## ORGANIC CHEMISTRY



Total Marks: 49

Max. Time: 50 min.

**Topic: General Organic Chemistry** 

#### Type of Questions

M.M., Min.

Single choice Objective ('-1' negative marking) Q.1 to Q.15

[45, 45]

- Subjective Questions ('-1' negative marking) Q.16
- (3 marks, 3 min.) (4 marks, 5 min.) [4, 5]
- 1. How many lone pairs of electrons are there in the given compound?

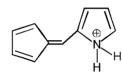
(D) 6

- (A) 4
- 2. How many lone pairs of electrons are there in the given compound?

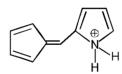
- (A) 4
- (B) 3
- (C) 8

(C) 8

- (D) 6
- 3. What is the hybridisation of positively charged nitrogen atom?

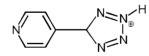


- (A) sp
- (B) sp<sup>2</sup>
- (C)  $sp^3$
- (D) None of these
- 4. How many sp<sup>2</sup> hybridised atoms are there in the given cation?

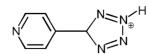


- (A) 8
- (B)9
- (C) 10
- (D) 12

5. How many lone pairs are present in the given cation?



- (A) 3
- (B) 4
- (C) 5
- (D) 6
- 6. How many N atoms are sp<sup>2</sup> hybridised in the given cation?



- (A) 3
- (B) 4
- (C) 5
- (D) 6

7. In which of the following molecules is the nitrogen atom sp<sup>2</sup> hybridised?

(A) 
$$HO_2C$$
 $OD_2H$ 
 $ODD_2$ 

(B)  $HO$ 
 $ODD_2$ 
 $ODD_2$ 

8. Geometry around how many carbon atoms is tetrahedral in the given structure?

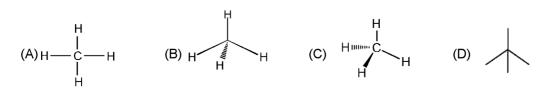
9. How many sp³ hybridised carbon atoms are there in the given anti-cancer compound (podophyllotoxin) ?

(A) 4

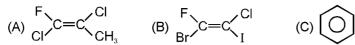
the anti-cancer compound podophyllotoxin

- 10. Bond order of C–C bond in benzene is:

  (A) 1 (B) 2 (C) 1.5 (D) two of above
- 11. Which of the following is correct three dimentional representation of  $CH_{4}$ ?







(B) 
$$F C = C C$$

In which of the following all carbon atoms are present in the same plane? 13.



14. In which of the following are all C atoms linearly arranged?

(A) 
$$CH_2 = CH - C = CH$$
 (B)  $CH = C - CH_2 - CH_3$  (C)  $CH_3 - C = C - CH_3$  (D)  $CH_3 - C = C - CH_3$ 

15. In 
$$\frac{CH_3}{CH_3}C = C = C \frac{CH_3}{CH_3}$$
 How many carbon atoms are linearly arranged?

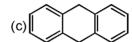
(A) 1

- (B) 8
- (C) 3
- (D) 7

16. A maximum of how many (i) atoms (ii) carbon atoms must lie in same plane in each of the following molecule?









No. of sp<sup>2</sup>–sp<sup>2</sup>

 $\sigma$ -bonds

No. of sp2-sp

 $\sigma$ -bonds

# **Answer Key**

### **DPP No. #10**

1. (D) 2.

(B)

3. (C) 4.

(C)

5. (B)

6. (C) 7. (C)

(C) 8.

9. (C) 10. (C)

(C) 11.

(A) 12.

(C) 13.

14. (D)

(C) 15.

16.

(b)

(f)

(i) 12, (ii) 8.

(c)

(i) 12, (ii) 8.

(d).

(i) 6, (ii) 4.

(e)

(a)

(i) 12, (ii) 6.

(i) 14, (ii) 8.

(i) 14, (ii) 9.

